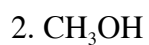


CHE 139
Activity 4

Determine the ideal angle and molecular geometry for each central atom in of the following molecules and ions. Use Lewis dot structures to help you. Most of these were on the last activity, so you can get the structures from there (if they were right...)



6. C_2H_2

7. NH_4^+

8. H_2CO

9. NO_2 treat a lone e^- like a lone pair

10. $NOBr$